

PITTING ON CARBON CATHODES IN ALUMINIUM ELECTROLYSIS CELLS

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Presented by: **Samuel Senanu**

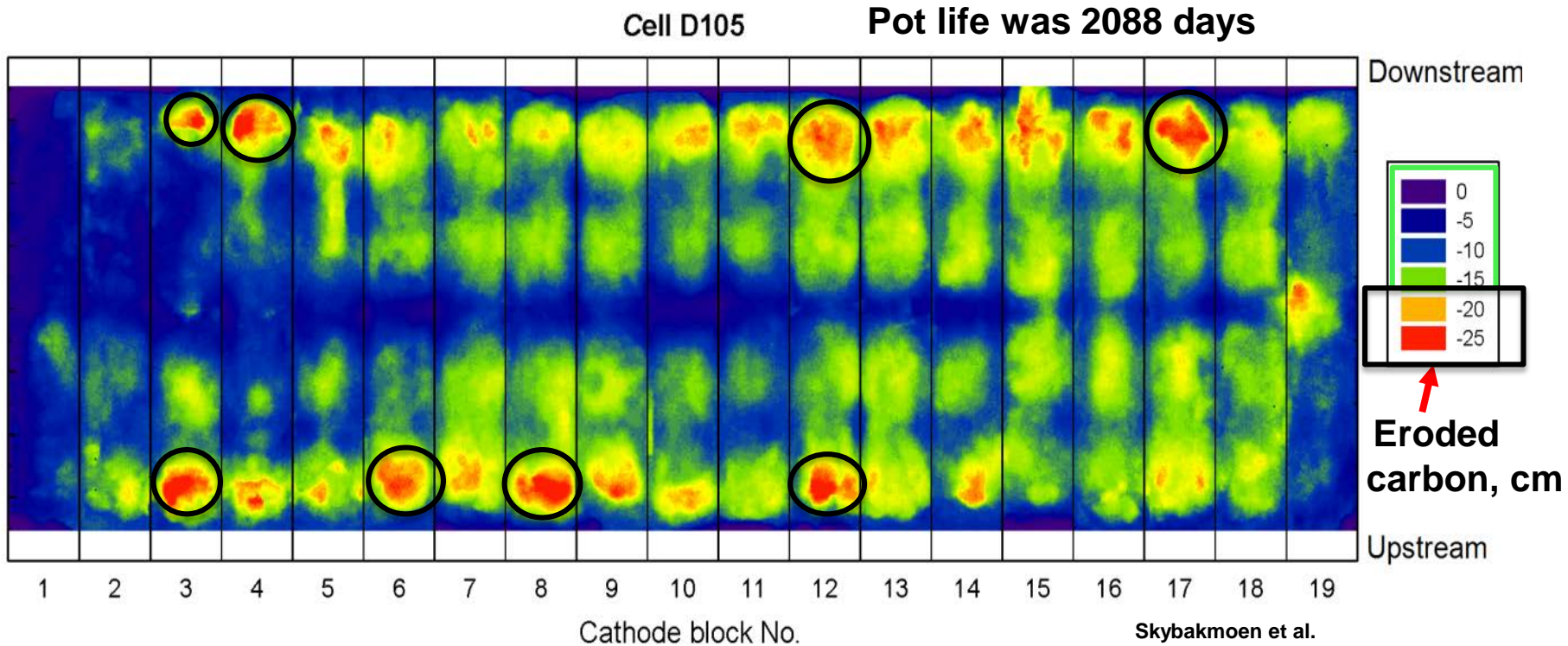


Belem, 29 October – 1 November 2018

Presenter's Bio

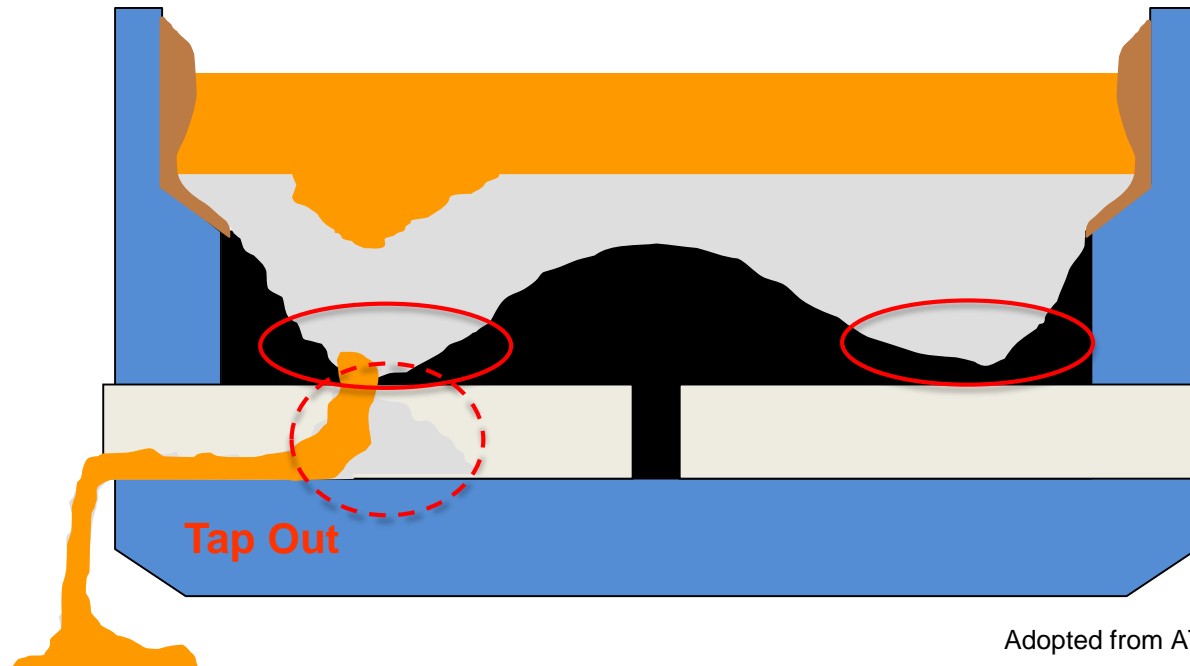
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- **Degrees and date earned: MSc Light Metals Production, August 2008.**
- **Affiliation: NTNU - Norwegian University of Science and Technology**
- **Present position: PhD Candidate**
- **Work experience: Process Engineer Potrooms, Alcoa Norway (2008-2016)**

Motivation



Motivation

A single pothole is enough to kill a cell!



Autopsy Analysis

- **Visual observations, profilometer and 3D Scanning**
 - **Topography of wear**
 - **Variation in wear patterns**

- **Micro analysis (SEM, EDS, XRD, CT Scanning)**
 - **Interface analysis**
 - **Chemical composition**
 - **Microstructure**

Overview of Autopsied SPL

Pot	Arrangement/ Technology	Amperage(kA)/ Cathode Current Density(A/cm ²)	Pot age (days)	Carbon Cathode Type	Wear Pattern	Comments
1	Side by Side/ Prebaked	313 / 0.8	2461	Graphitized and Impregnated	WW	Planned shutdown
2	End to End/ Prebaked	175 / 0.8	1028	Graphitic (100% Graphite agg.)	W	Tapout
3	End to End/ Prebaked	175 / 0.8	3154	Graphitic (100% Graphite agg.)	W	Planned shutdown
4	End to End/ Prebaked	175 / 0.8	2849	Graphitic (100% Graphite agg.)	W	Planned shutdown
5	Side by Side/ Prebaked	313 / 0.8	1731	Graphitized and High Density	WW	Tapout

What was Done



Size of the Spent potlining = 4 m X 15 m

Wear pattern measurements

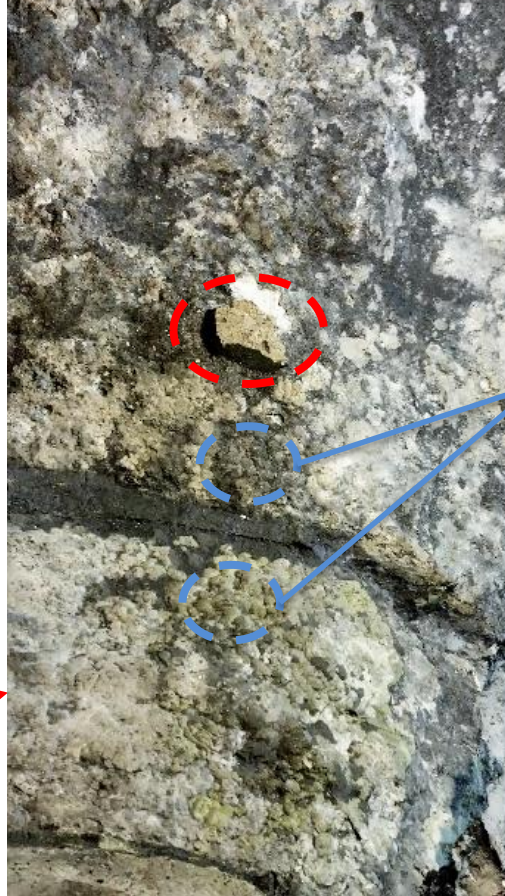


Photography



Sample Collection

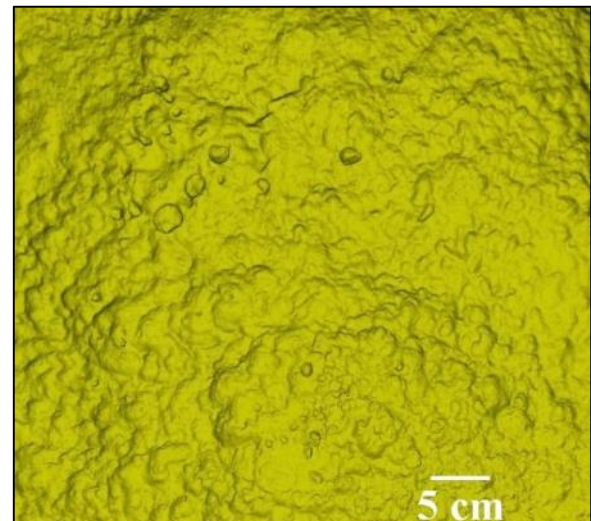
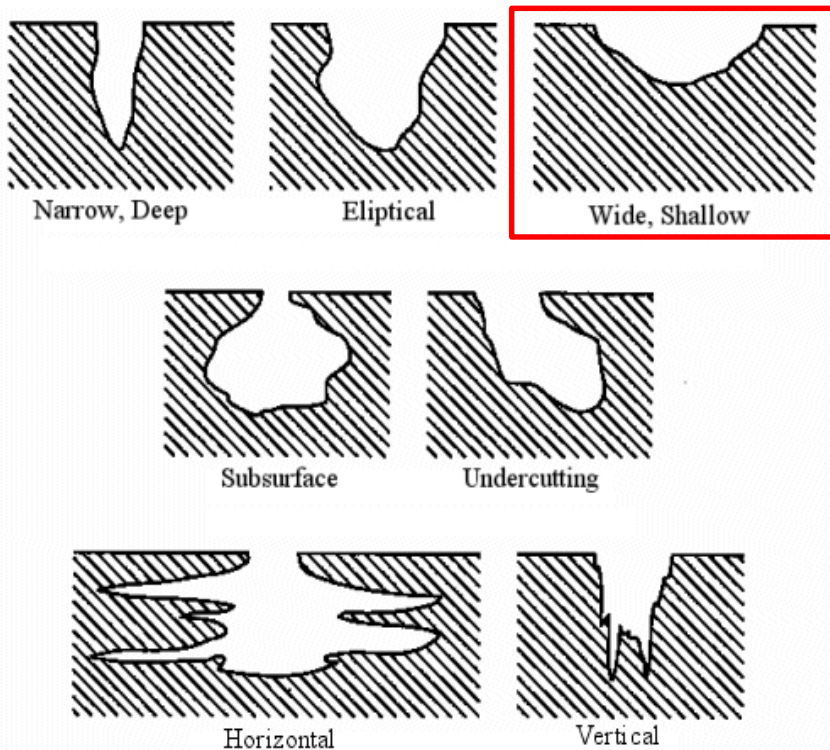
Pitting on Spent Potlining



**Pothole location
characterized
by pitting**

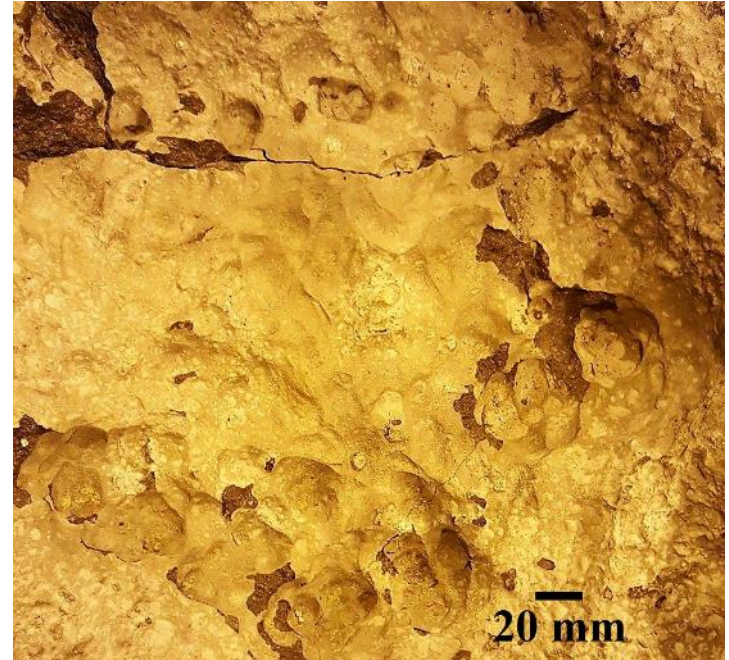
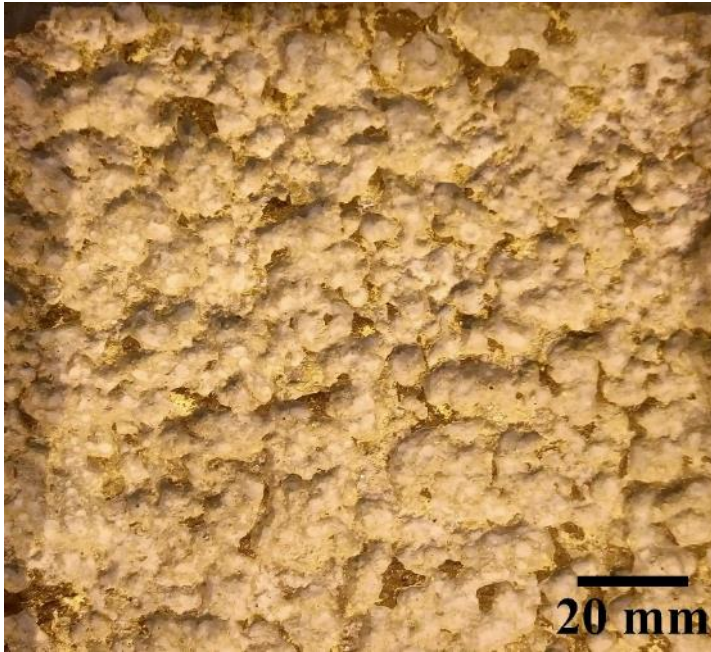
Pitting

Pitting is a term adopted from pitting corrosion of steel which is a highly localized form of corrosion that produces sharply defined cavities

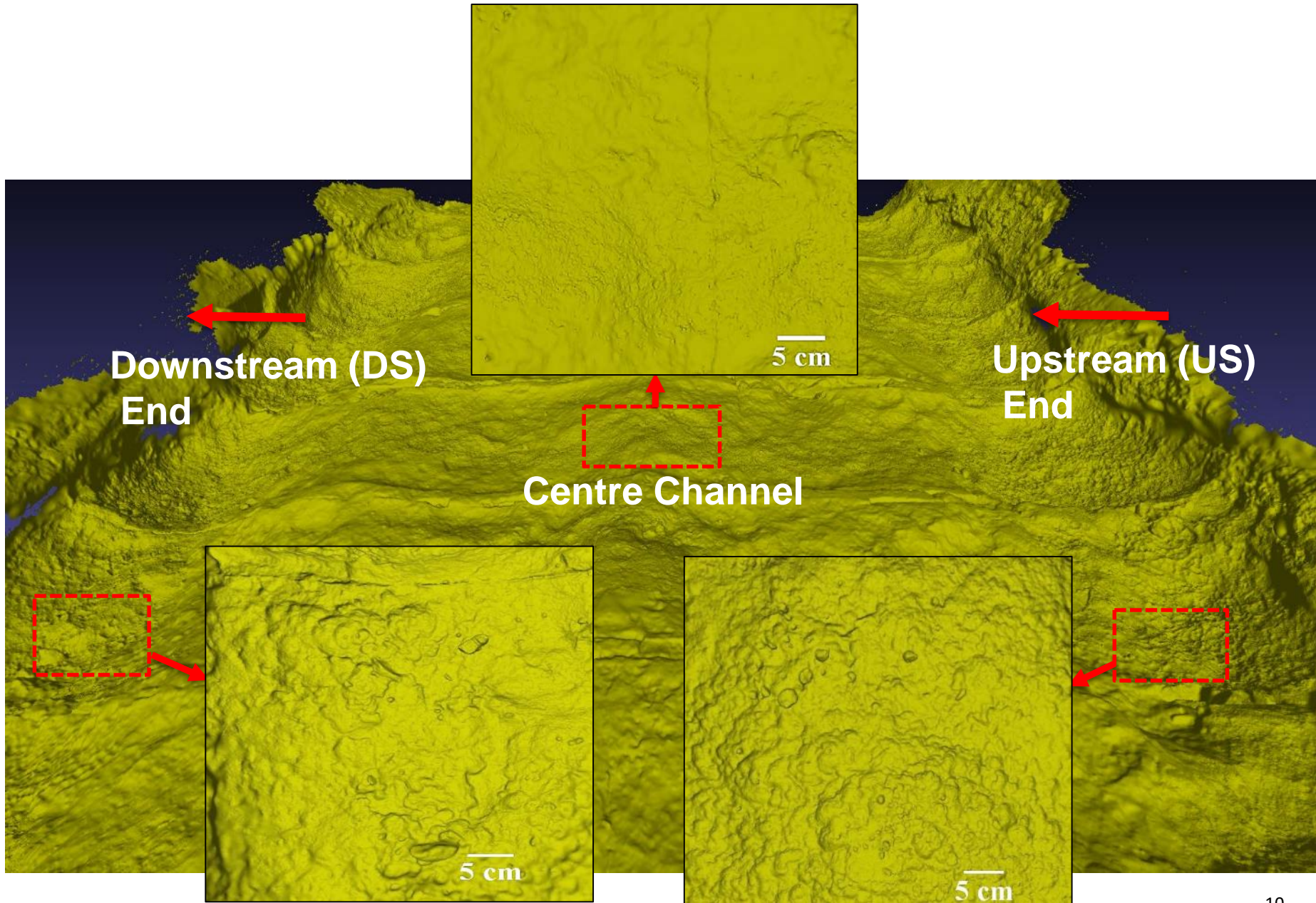


Ref: J.R. Davis, *Corrosion*:

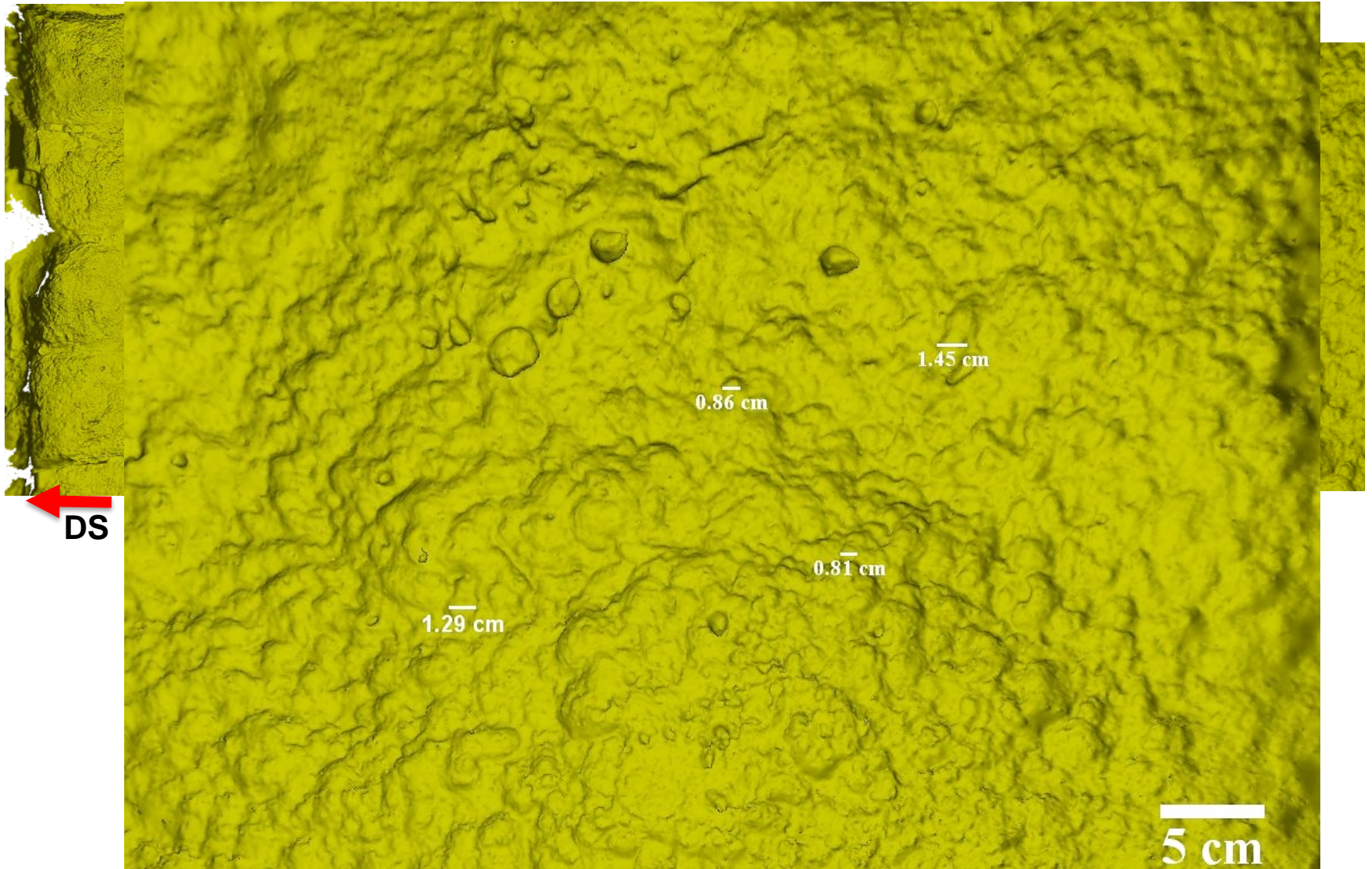
Pitting on Spent Potlining



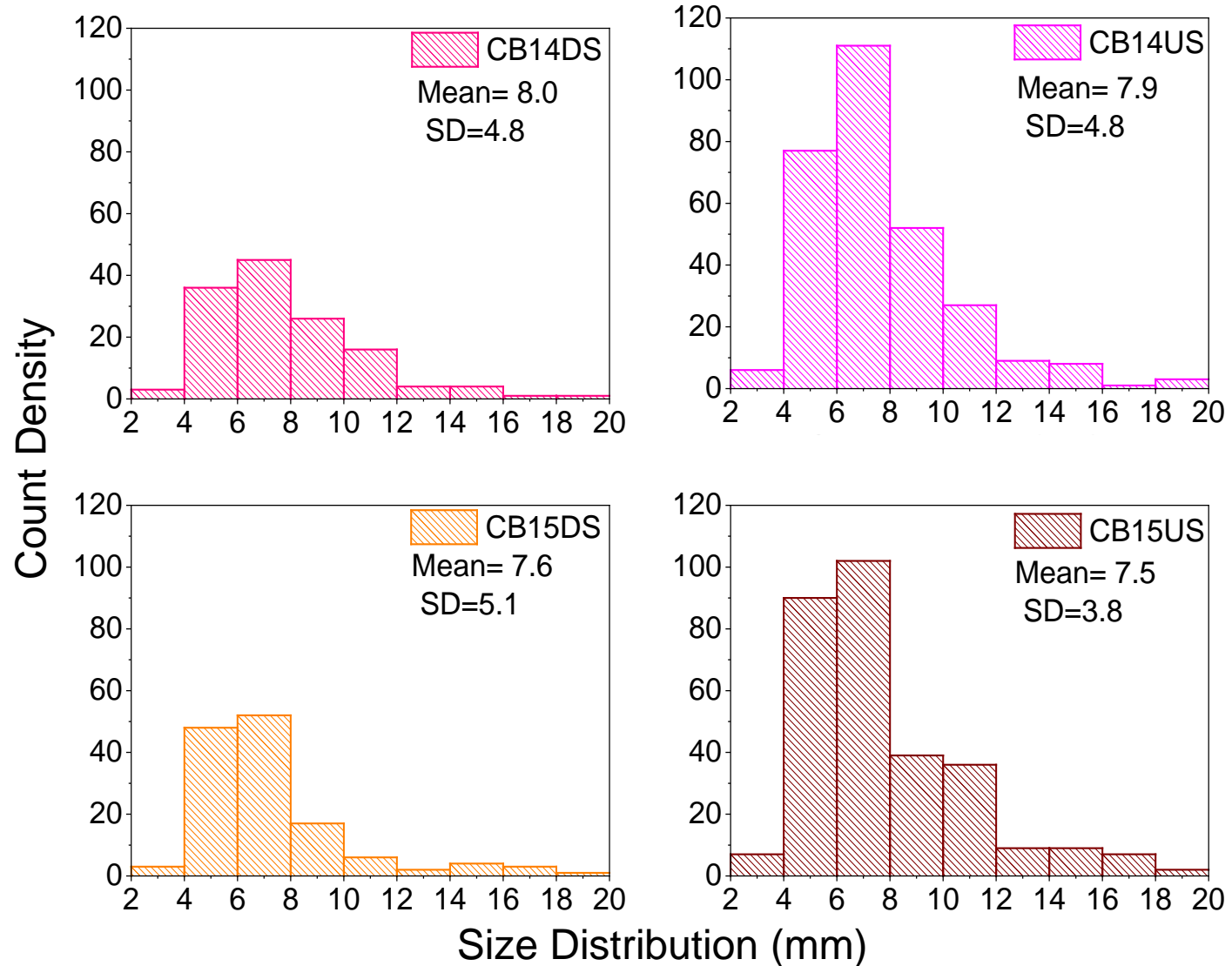
Pitting on Spent Potlining (3D scanning)



Characterization of Pitting



Characterization of Pitting



Microstructure Analysis

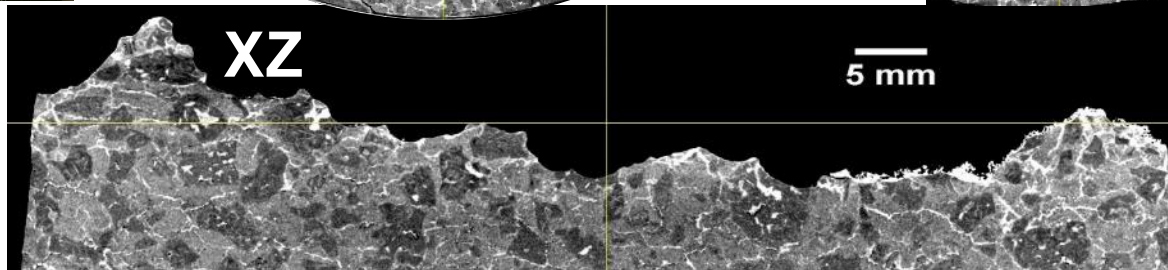
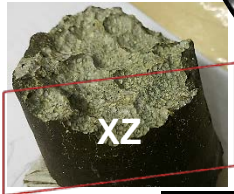
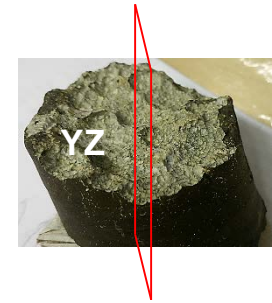
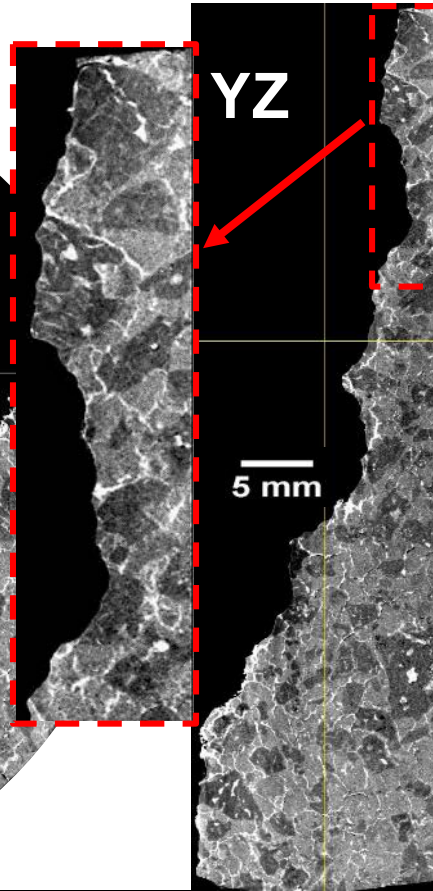
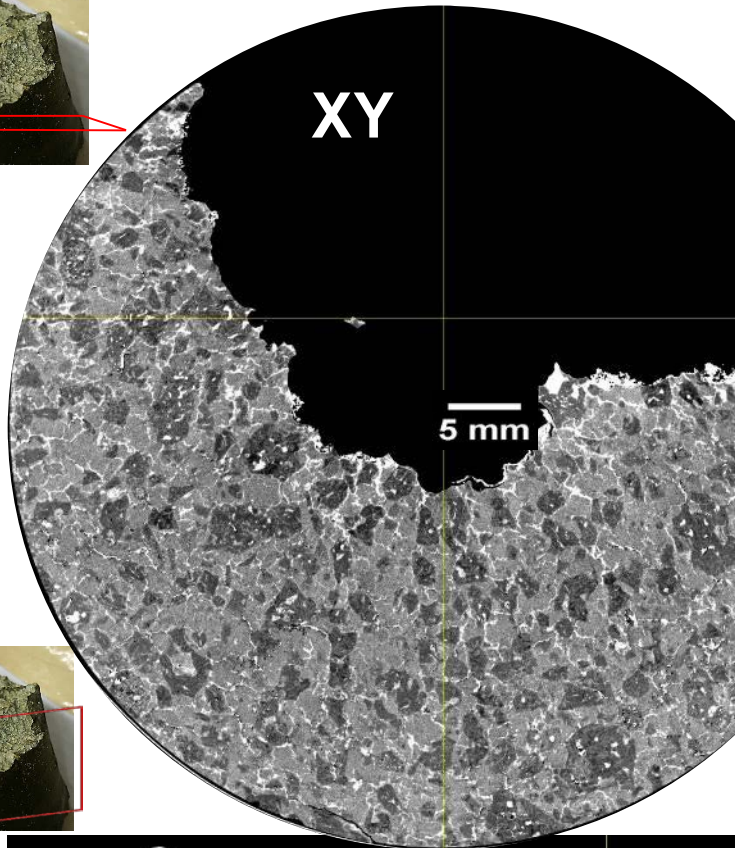
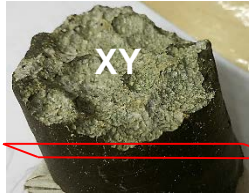


Samples for CT analysis



Samples for Microscopy analysis

Microstructure Analysis (CT)



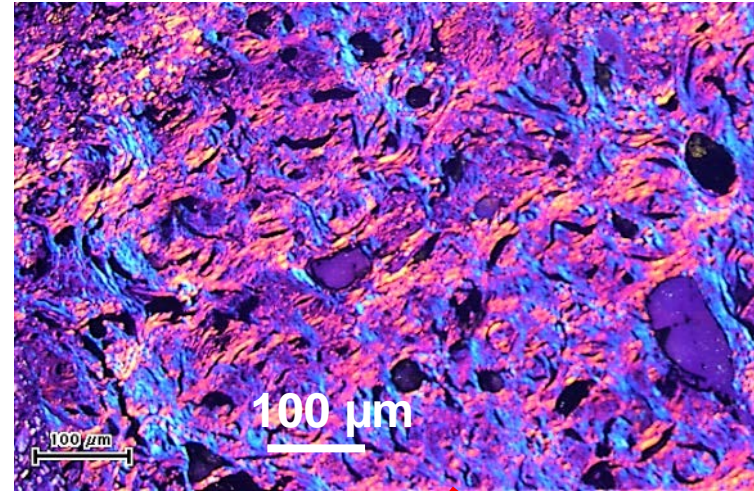
**Remains of coke
Aggregate within
carbon
Matrix**

**Relatively uniform
wear
of aggregate and
binder
matrix**

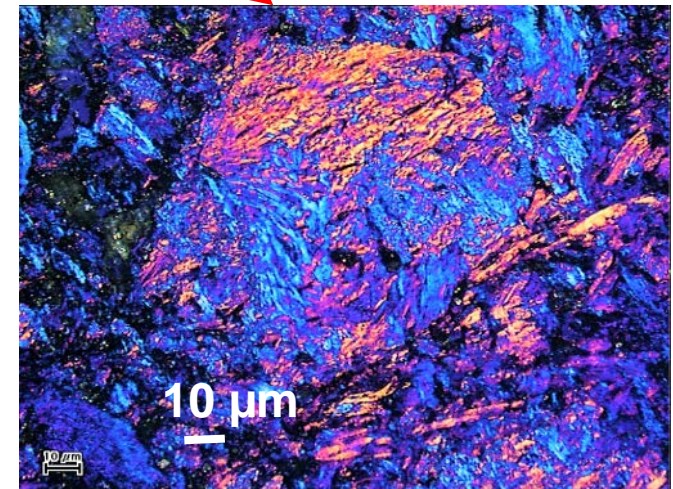
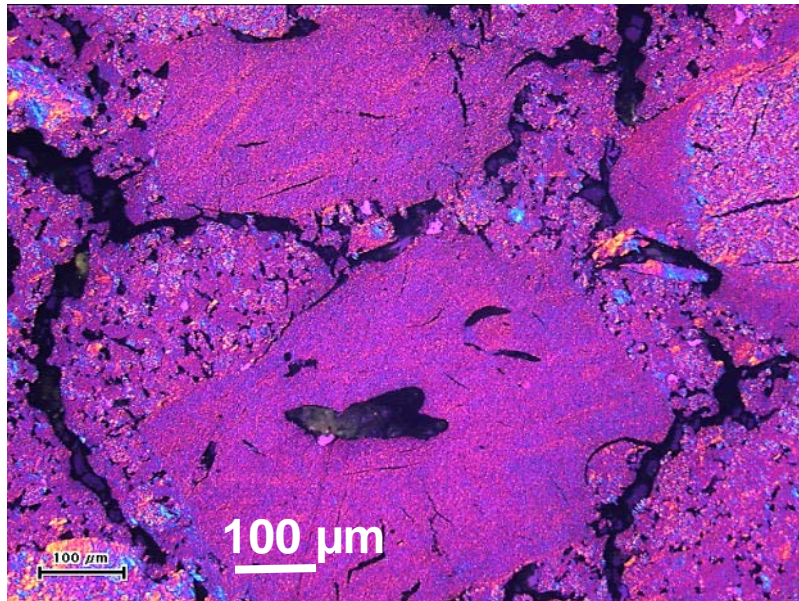
Microstructure Analysis



Graphitic Cathode block

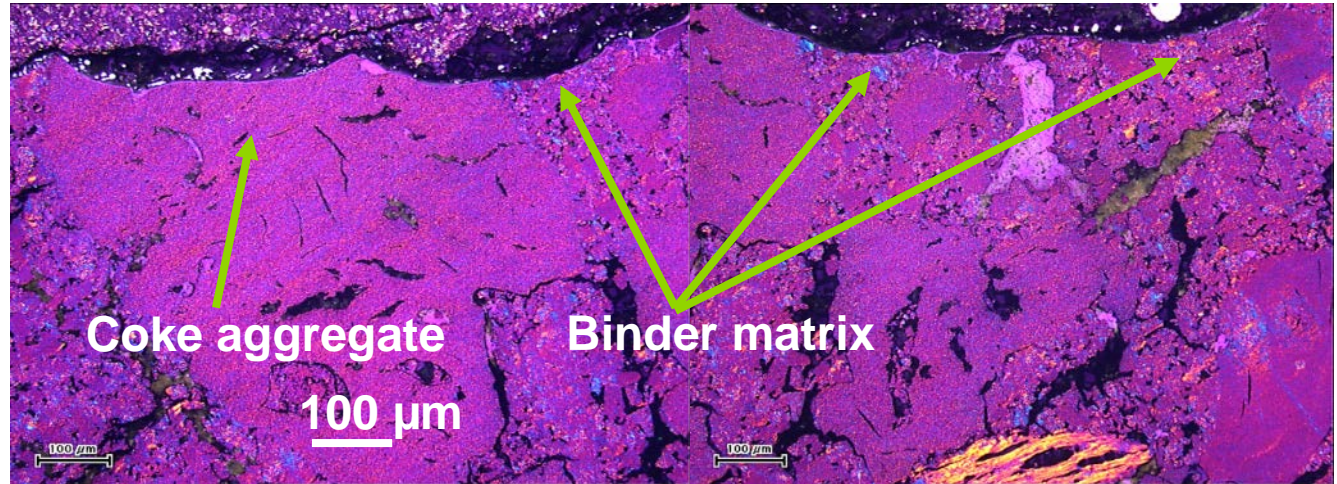
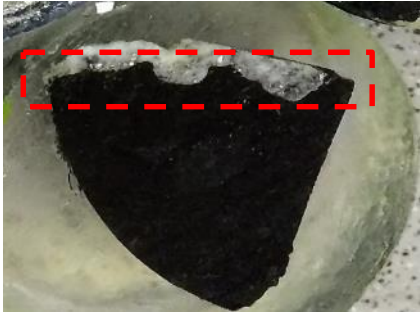


Graphitized Cathode block

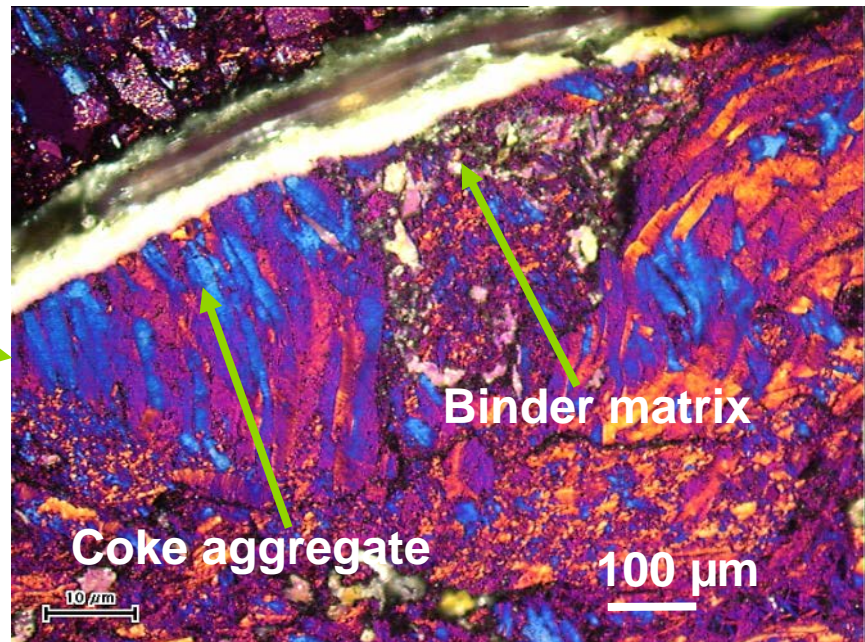
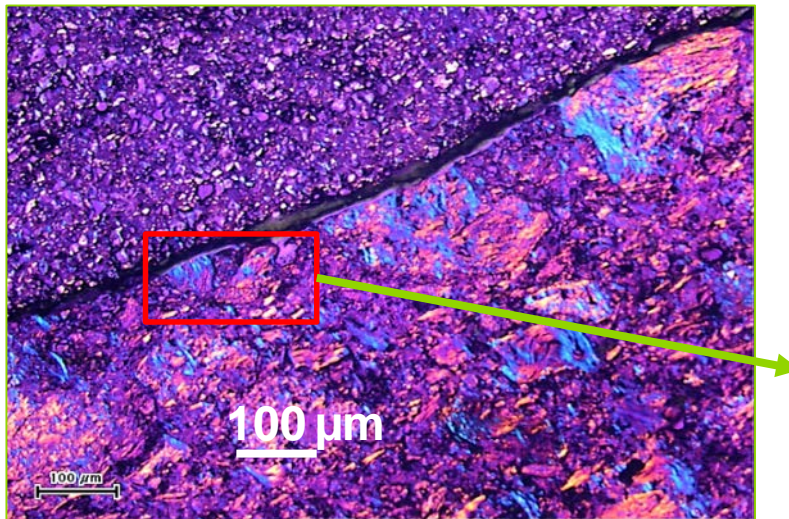


Microstructure Analysis of Pitting

Graphitized Cathode block



Graphitic Cathode block



Summary of Autopsy Results

- **Highest wear at side ends of carbon cathode blocks**
- **Locations with highest wear characterized by high degree of distinct pitting**
- **Graphitized cathode blocks with very low degree of pitting at centre canal**
- **Pitting observed all over graphitic cathode block surface**
- **Variations in size of pitting from centre canal to side end of cathode blocks**
- **Remains of worn out coke aggregate observed in carbon matrix**

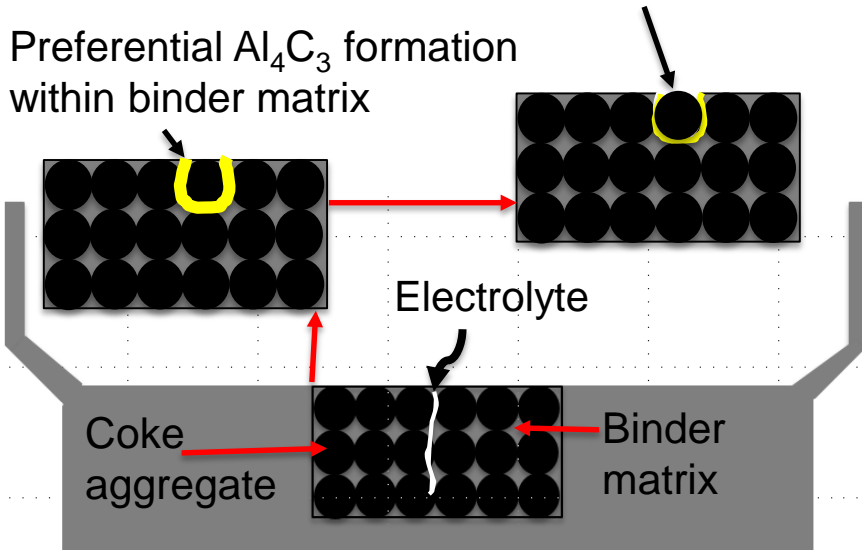
Discussions

What is the mechanism behind pitting formation on carbon cathode blocks?

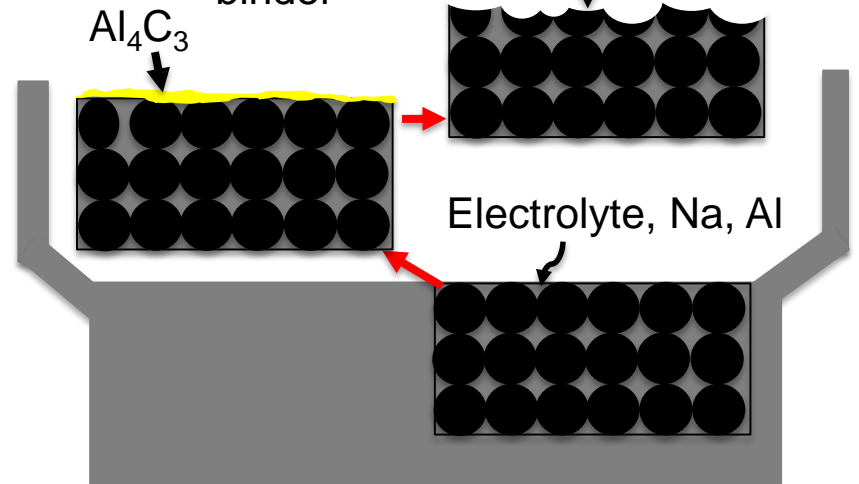
- Aggregate grain detachment due to weakened binder matrix
- Electrochemical / Chemical wear of carbon cathode (binder and aggregate)

Aggregate detachment due to weakened binder phase

Preferential Al_4C_3 formation within binder matrix

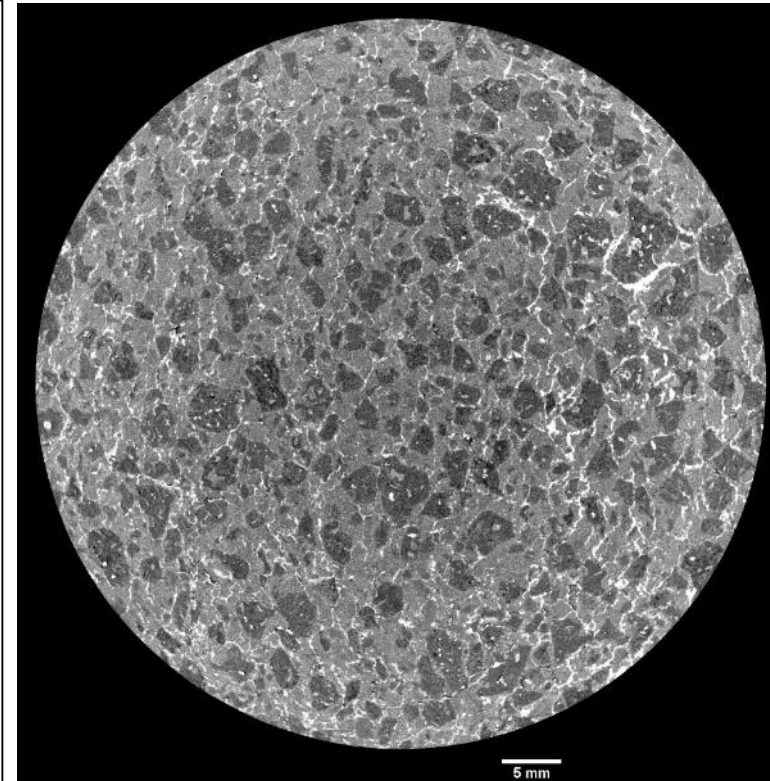
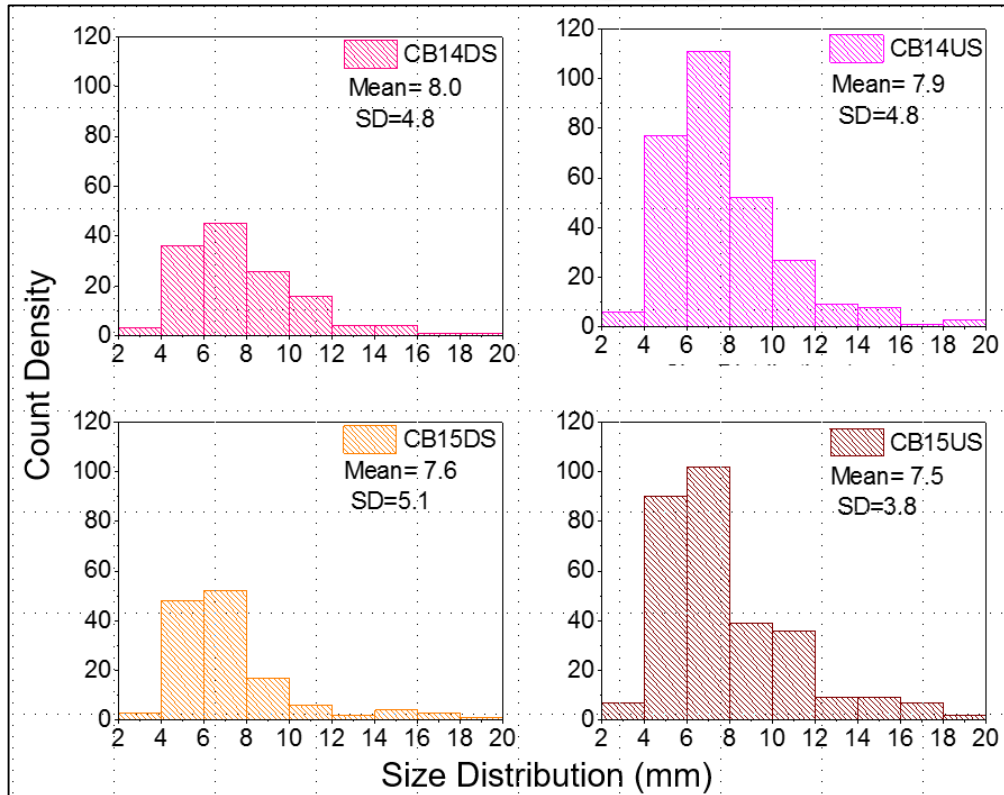
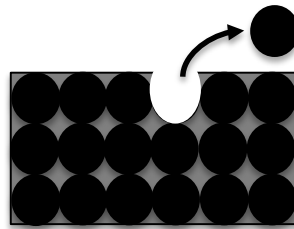


Dissolution and transport of Al_4C_3 leading to wear of aggregate and binder



Discussions

Verify grain detachment mechanism

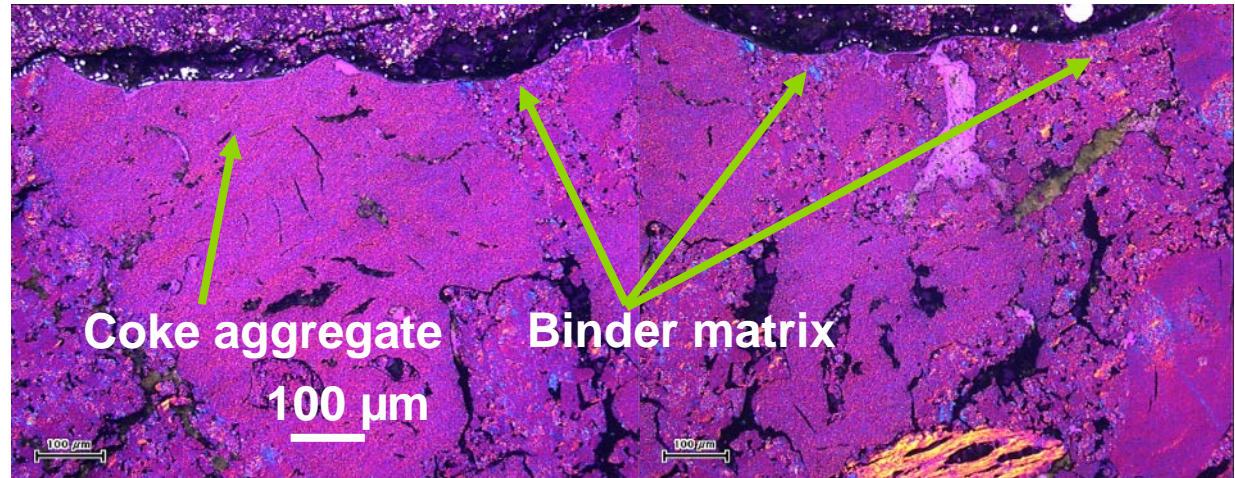
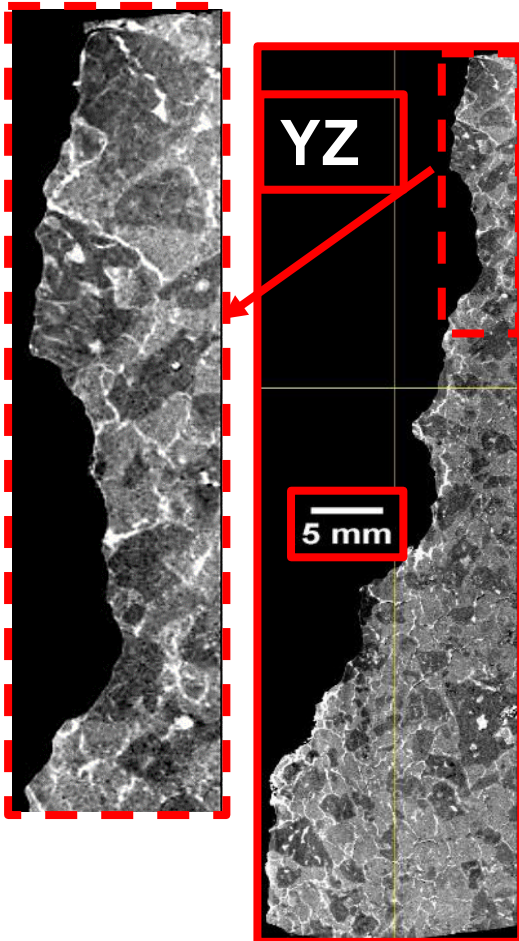


Average pitting size = **7.4 mm**

Average aggregate grain size = **3 mm**

No correlation between aggregate grain and pitting size \Rightarrow No grain detachment

Discussions



Remains of coke aggregate within carbon matrix as shown by CT image and optical micrographs questions particle detachment as a possible mechanism

A relatively uniform wear front suggest a chemical or electrochemical mechanism

Conclusions

- **Pitting occurs on all carbon cathode types**
- **Current density and faster transport rates might be important for pitting formation and development**
- **Occurrence of pitting at locations with highest wear might suggest it as a precursor for cathode wear**
- **Non-correlation between pitting and aggregate size and presence of aggregate remains with carbon matrix questions particle detachment as possible mechanism for pitting formation**
- **A relatively uniform wear front suggest chemical or electrochemical mechanism**

Acknowledgement

Financial support from the Norwegian Research Council and the partners Hydro, Alcoa, Elkem Carbon and Skamol through the project "Reactivity of Carbon and Refractory Materials used in Metal Production Technology" (CARMA) is acknowledged.



Belem, 29 October – 1 November 2018